PvBOP® 1 AND PyBroP: TWO REAGENTS FOR THE DIFFICULT COUPLING OF THE α , α -DIALKYL AMINO ACID. Aib.

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Summary The difficult coupling of α -aminoisobutync acid (Aib) was carried out using PyBOP® and PyBroP in a comparative study with BOP and BroP These reagents gave good results under simple conditions (one pot, rt, 1 h) Coded amino acids could be coupled with Aib using PyBOP under standard conditions of peptide synthesis without racemization whereas the coupling of two Aib residues required PyBroP/DMAP A fragment containing an Aib C-terminal could be coupled without epimerization of the penultimate residue

Introduction

Aib (α -aminoisobutyric acid) is a noncoded amino acid present in peptabol antibiotics, which are isolated from fungi² and have a capacity to form voltage-dependent ion channels in bilayer membranes³ The introduction of Aib into a peptide can result in modifications that are of interest conformationally (formation of α or 3₁₀ helices, or ß-turns)⁴ and pharmacologically⁵ However, the introduction of Aib into a peptide remains difficult⁶⁻¹¹ The classic reagent DCC (or DCC/HOBt) gives mediocre yields despite long reaction times^{12,13} Other methods have been used, i.e. mixed anhydrides^{6,7}, active esters^{12,14}, Aib oxazolone⁷, but none of them gives consistently satisfactory results Coupling reagents, such as 2-chloro 4,6-dimethoxy-1,3,5-triazine¹⁵ or diethyl phosphorobromidate¹⁶ have been used in rare cases

An original method using an azirine as the synthetic equivalent of Aib has recently been α developed¹¹ Specifically, it amounts to coupling with the Aib amine, which, after two steps, produces a peptide having an acidic Aib C-terminal Further coupling with this carboxyl must be carried out using classical methods¹¹ involving the limitations described above

Because of the difficulty of this coupling and the presence of many Aib residues in peptaibol sequences, these compounds have only rarely been synthesized, and in these cases in solution^{13,17,18} For the same reasons, SPPS (solid-phase peptide synthesis) is unsuitable, although it has been used to obtain alamethicin¹⁰

We recently developed new peptide coupling reagents of the BOP¹⁹ family PyBOP[®] 2, BroP 3, and PyBroP 4. PyBOP20 2 gives results that are at least as good as BOP, and it avoids the use and formation of carcinogenic HMPA BroP²¹ 3 et PyBroP²² 4 give better results than reagents 1 and 2 when coupling Nmethylated amino acids The results we have obtained when coupling N-methylated amino acids with BroP and PyBroP, as well as the corresponding differences in reactivity observed between reagents 1 and 2, containing the oxybenzotriazole residue, and the brominated reagents 3 and 4, led us to study their behavior during Aib coupling In the present paper, we report results obtained using reagents 1 to 4 in Aib coupling during the synthesis of di- and tripeptides

Results and Discussion

I/Yields

In a first series of reactions (table I, compounds 5-11), the four reagents 1 to 4 gave high yields (80 to 95% in 1 h of reaction) even with hindered residues (compounds 6-11) The results were identical whether A1b was in the C- or N-terminal position Moreover, there were no differences among the four reagents A second series of reactions (compounds 12-15) involved the difficult coupling of two Aib residues The decrease in reactivity is particularly important with the bulky Boc protective group using

E. FRÉROT et al.

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BroP and PyBroP (compound 13 25%, 24 h) However, with the two latter reagents, catalysis with DMAP

made it possible to obtain good yields in only 1 h of reaction, although synthesis of Boc-Aib-Aib-OBzl 15 remained difficult even with DMAP catalysis For coupling of Z-Aib with a very weakly nucleophilic amme, benzocame (p-NH₂-Ph-COOEt), the use of BroP(PyBroP)/DMAP made it possible to obtain 66% (65%) armde 16 m 2 h

Comparison of these results with those in the literature shows that reagents 1-4 are clearly more effective than DCC with respect to yield (compound 5), and reacnon ume (compounds 8,12,15) Compared to active esters in the presence of imidazole, the results are either similar (compound 6) or better (compounds 8,12) The same observation 1s also true for rmxed anhydnde methods (compounds 9,12,16 and 13) With regard to other peptide coupling reagents diethyl phosphobromidate¹⁶ gives results (compound 12 85%, 2 h) comparable to those we obtained Dipeptide 15 could not be obtained with BOP- $Cl²³$, which was also true of CDI in the case of compound 16 ²⁴ Nevertheless, our results were not as good as those obtained by the azirine method For example, Z-Alb-Alb-OH is obtained in two steps with 91% yield²⁵ whereas PyBroP/DMAP yielded 80% of the corresponding methyl ester 12 Similarly, Z-Val-Aib-OH is obtained with 95% yield¹¹ as opposed to 90% Z-Val-Alb-OMe 7 with BroP It should be noted, however, that in the azirine method, the precursor has to be prepared beforehand

The reagents we tested showed certain advantages Compounds 2, 3 and 4 are, like BOP, stable solids that are easy to handle Moreover, PyBOP and PyBroP are now commercially available 26 The reactions can be carried out one pot, at room temperature C-protected amino acids can be coupled in the form of a free amine (compound 8), chlorhydrate (compounds $5, 6, 7$, etc), tosylate (compound 15), or $trifluoroacetate (compound 18)$ In the case of BroP (and BOP) HMPA is eliminated during the work-up in the acid washes of the reaction mixture With PyBOP and PyBroP, tns(pyrrolldmo)phosphme oxide remaining in the crude product is easily eliminated by rapid chromatography on silica gel

II/ Epimenzation

It has been demonstrated that when a urethane-protected amino acid is coupled with $PyBOP^{20}$, there 1s no eplmenzatlon Moreover, when N-methylated ammo acids are coupled with BroP and PyBroP, epimenzation is undetectable by ¹H-NMR at 360 MHz^{21,22,27}

In the case of A1b, coupling with PyBOP also occurred without epimerization (Table II) The absence of diastereoisomer 24 (eq 3) during the synthesis of tripeptide 20 (eq 1), as measured by HPLC, showed that the coupling of Z-L-Leu with Aib-OAllyl (eq 1, compound 18) did not involve racemization Similarly, the chirality of Z-L-Leu was preserved during coupling with Alb-Pro-OtBu (eq 2, compound 20)

On the other hand, slight epimerization did occur when coupling with BroP and PyBroP, resulting in compounds 18 (eq. 1) and 20 (eq. 2) (Table II). Moreover, epimerization was measurable by ¹H-NMR

during the coupling of Z-Ile with Aib-benzocaine (eq 4, compound 26) Signals related to α protons of L-Ile and the epimerization product D-allo-Ile are separate²⁸. Thus, epimerization was evaluated at 0.5% with BroP and was not detectable with PyBroP

We considered it of interest to study the coupling of a peptide fragment containing an Aib at the Cterminal position In peptide synthesis, fragment coupling that extends the peptide in the direction $N \rightarrow C$

TABLE II % EPIMER BY STEP TO STEP COUPLING

*not detectable on HPLC

**not detectable on ¹H NMR

Equation 1

Equation 3

Equation 2

 $Z-D-Lcu + Ab-OMe \xrightarrow{BroP} Z-D-Lcu-Aib-OMe \xrightarrow{HO/H_2O} Z-D-Lcu-Aib-OH$ Z-D-Leu-Aib-Pro-OtBu Pro-OtBu 23 22 24

BroF $2-$ Aib-p-NH₂-CH₂-COOEt
H₂/Pd/C
HCl, Aib-p-NH₂-CH₂-COOEt
(PyBroP)

 25
 25
 25
 26

is used only rarely (except with Gly and Pro C-terminals) because it results in epimerization in the intermediate oxazolone (scheme 1 $R^3 = H$)

If an Aib carboxylate is activated (Scheme 1 $R^2 = R^3 = CH_3$), this epimerization mechanism does not occur and fragment coupling would appear to be advantageous since it allows convergent syntheses However, Heimgartner²⁸ and Bruckner²⁹ have shown that in this case the penultimate amino acid may be epimerized Heimgartner obtained as much as 50% epimer when coupling Z-IIe-Aib with benzocaine using

DCC/HOBt Chirality was preserved when the reaction was carried out in the presence of 10camphorsulfonic acid at 0°C Bruckner proposed a rule he called the "do's and don'ts of Aib peptidecoupling" the coupling of a P-Xxx-Aib fragment is formally forbidden if the chirality of the amino acid Xxx is to be preserved

We studied the coupling of the fragment Z-Leu-Aib with Pro-OtBu (eq 1), a group present in alamethicin This coupling falls into the category of "don'ts" in Bruckner's rule This author obtained as

TABLE III Z-Leu-Aib + Pro-OtBu COUPLING

*not detectable on HPLC

much as 49 3% epimer while preparing the oxazolone of peptides terminated by Leu-Aib²⁹, which shows the capacity of the Leu fragment to epimerize according to the mechanism considered here

The results in Table III show that epimerization was not detectable under any of the conditions studied This was verified by HPLC in comparison with the diastereoisomer Z-D-Leu-Aib-Pro-OtBu 24 The absence of epimerization with PyBOP and PyBroP (even in the presence of DMAP) led us to carry out the same coupling with DCC and DCC/HOBt Here again, the chirality of the Leu residue was preserved These results are in agreement with the low percentages, $1e$ 07% D-Leu

and 0 6% D-Val, present in the alamethicin synthesized by Schmitt and Jung¹⁸ who coupled Aib-terminated fragments with these amino acids in the penultimate position Although it is likely that oxazolone is involved during fragment coupling with Aib C-terminal activation^{17,24}, we do not consider that epimerization of the penultimate residue can occur under the usual conditions of peptide coupling

III/ Mechanism

The very difficult couplings (compounds 12, 13, 15, 16) are of particular interest in terms of the mechanism involved In the cases of Z- and Boc-Aib couplings with Aib-OMe (compounds 12,13) using BOP or PyBOP, thin layer chromatography revealed the immediate formation of an intermediate that slowly disappeared in favor of the dipeptide This kind of intermediate was obtained (TLC) by the action of PyBOP on Boc-Aib Its ¹H-NMR spectrum showed it to be the oxybenzotriazole ester of Boc-Aib 14 During the coupling of Z-Aib with benzocaine (compound 16), the oxybenzotriazole ester of Z-Aib was also formed It reacted even more slowly because of the low nucleophilicity of the amine

In the general case of BOP coupling, we have proposed 30 the following mechanism, an acyloxyphosphonium salt is first formed, which is aminolyzed by the C-protected amino acid (Scheme 2,

pathway a) The results given above show that in the case of Aib, if the steric hindrance is large (compounds 12, 13) or the amme is weakly nucleophilic (compound 16), the amon -OBt reacts with the acyloxyphosphonum salt to produce the oxybenzotriazole ester (Scheme 2, pathway b) The latter then reacts slowly to yreld the expected product This result is perfectly comparable wtth those obtamed in the coupling of two hindered N-methylated amino acids²¹

For difficult couphngs of Arb with BroP and PyBroP, a pathway via a symmemc anhydrrde can be considered (Scheme 2, pathway c) These reacttons are markedly accelerated by DMAP, a reagent known to catalyze the amtnolysts of anhydrtdes (It has also been shown that BroP can be used to prepare anhydrides³¹) This result differs from those obtained^{21,22} with hindered N-methylated amino acids for which DMAP catalysis was unnecessary

Conclusion:

The results obtained and the practical advantages reported here indicate that reagents 1-4 are very suitable for Aib coupling No difference was found between BOP and PyBOP or BroP and PyBroP However, in view of the toxicity of HMPA, PyBOP and PyBroP may be preferable. PyBOP provides good results in most cases and produces peptides that are free of epimenzation In the case of difficult coupling mvolvmg two Arb residues, the reaction 1s slow wnh PyBOP, but tt can be camed out with PyBroP m the presence of DMAP The results obtained with these two reagents suggest that they can be used in sohdphase synthesis of peptarbols Studies on this subject are now in progress m our laboratory

Experimental Section

Melting points are uncorrected ¹H NMR spectra were recorded on a Bruker WM-360 instrument at 360 MHz Optical rotations were taken at 20°C on a Schmidt & Haensch Polartronic D apparatus and are at \pm 1° Elemental analyses were obtained from the "Service Central d'Analyse du CNRS" FAB mass measurements were done by the "Laboratorre de mesures physiques". USTL, Montpelher TLC were performed on silica gel GF254 aluminium sheets (0 2 mm thick; Merck) Column chromatographies were performed using sthca gel (0 063-O 200 mm, Merck) Analyucal HPLC was carrted out on a Beckman System Gold When $1H$ NMR and elemental analyses are not given, the products are identical to those prevrously described m the hterature

Coupling methods

BOP, PyBOP, BroP **and** PyBroP. **General procedure: 2** mm01 of DJEA (3 mmol If amme salt was present) were added to a solution of 1 mmol N-protected acid component , 1 1 mmol of the C-protected ammo acid (or amine salt) and 1 mmol coupling reagent in 1 ml of CH_2Cl_2 (filtered on Alumina) The reaction was stirred at r t for 1 h (except where otherwise stated), evaporated, redissolved in 20 ml of EtOAc and washed 3 x with 5% KHSO₄, brine, 3 x 5% NaHCO₃, brine, dried on Na₂SO₄, filtered and evaporated All dipeptides obtained were purified by column chromatography on silica gel using hexane/EtOAc (except when racemization was studied)

Boc-Aib was synthesized from Aib (3 g) with 3 eq Boc-F (tert-butyl fluorocarbonate, 50%wt in monoglyme, a gft from Propephde) in 1N NaOH/dloxan (100 ml) at pH=9,5 Dloxan was removed *m* vacuo, the solution acidified, and the product extracted from EtOAc, washed 3x with brine, dried on Na₂SO₄, evaporated Yield 69% M p = 121-122 °C (lit ¹² m p = 118 °C) Rf = 0 4 (hexane EtOAc, 50 50 l%AcOH)

HCl.H-Aib-OMe was synthesized according to Boissonnas et al ²⁹ From Aib (4 g, 39mmol) Yield 93%. M p = 179-182°C (ht $12 \text{ m p} = 179-183$ °C)

Tos.H-Aib-OBzl: A solution of Aib $(2 g, 19 mmol)$, APTS $H₂O$ $(3 75 g, 19 mmol)$ and benzylalcohol (8 ml) in toluene (20 ml) was azeotropically refluxed for 16 h. The colourless solid was filtered and washed with ether Yield 78% M p = $152-153\textdegree C$ (ht ¹² m p = $154-156\textdegree C$) **Z-Aib-Gly-OEt (5).**

with BGP, Z-Am-OH (0 474 g, 2 mmol), HCl Gly-OEt (0 307 g, 2 2 mmol), (0 884 g, 2 mmol) BOP and DIEA (104 ml, 6 mmol) in CH_2Cl_2 (2 ml) Column chromato (92%). Oil Rf = 0 5 (hexane EtOAc, 1 1) ¹H NMR (CDCl₃) δ aphy (hexane EtOAc, 1 1) Yield 0 57 g 1 25 (3H, t, J = 7 1 **HZ, 0** CH2C_3), 152 (6H, s, CH3 A1b), 3 97 (2H, d, J = 4 9 Hz, CH2 Gly), 4 18 (2H, q, J = 4 1 Hz, OCH2CH3), 5 07 (2H, s, CH2 Z), 5 35 (1H, s, NH Aib), 6 79 (1H, broad s, NH Gly), 7 26-7 35 (5H, m, C₆H₅ 2) Anal calc for $\rm{C_{16}H_{22}N_2O}$ $1/2$ H₂O %C = 57 99, %H = 7 00, %N = 8 45 Found %C = 58 00, %H = 6 85, %N = 8 53

 $\frac{\mathbf{W} \cdot \mathbf{U}}{\mathbf{H}}$ From Z-A1b-OH (2 mmol) Yield 87%

With BroP, From Z-Aib-OH (0.5 mmol) Yield 89%

With $PvBrOP$, From Z-Aib-OH (4 mmol) Yield 87%

Z-Aib-Val-OMe (6):

With BOP: Z-Alb-OH (0 237 g, 1 mmol), *HCl H-Val-OMe (0 184 g, 1 1 mmol*), BOP (0 442 g, 1 mmol) and DIEA (0 52 ml, 3 mmol) in CH₂Cl₂ (1 ml) Column chromatography (hexane EtOAc, 60 40) Yield 0 294 g (88%) M p = 74°C ($\ln 12$ m p = 100-102°C 1) $\left[\alpha\right]_{D}$ = -3° (c=1, EtOH) Rf = 0 55 (hexane EtOAc, 60 40) ¹H NMR (CDCl₃) 8 0 83 (3H, d, J = 6 8 Hz, CH₃ Val), 0 90 (3H, d, J = 6 4 Hz, CH₃ Val), 1 5 (3H, s, CH₃ Alb), 1 55 (3H, s, CH₃ Alb), 2 09-2 18 (1H, m, H-B Val), 3 70 (3H, s, OCH₃), 4 99 (1H, dd, 3 = $49 \text{ Hz}, \text{ J} = 8.8 \text{ Hz}, \text{ H-}\alpha \text{ Val}), 5 \text{ 06-5 10 (2H, AB, J = 12 2 Hz, CH}_2 \text{ Z}), 5 \text{ 28 (1H, broad s, NH Aib)}, 6 \text{ 79}$ (1H, broad, NH Val), 7 27-7 37 (5H, m, C_6H_5 Z) Anal calc for C_1 %N = 7 99. Found %C =61 90, %H =7 33, %N = 7 79 $_{18}H_{26}N_{2}O_{5}$ %C = 61 70, %H = 7 48,

With BroP, From Z-Atb-OH (0 5 mmol) Yield 87 %

With PyBroP; From Z-Aib-OH (2 mmol) Yield 84%

Z-Val-Aib-OMe (7):

With BroP, Z-Val-OH (0 251 g, 1 mmol), HCl H-Alb-OMe (0 151 g, 1 1 mmol), BroP (0 388 g, 1 mmo]) and DIEA (0.52 ml, 3 mmol) in $CH₂Cl₂$ (1) 0 31 g (90%) M p = 83-84°C $[\alpha]_D = 24^\circ$ ml) Column chromatography (hexane EtGAc, 60 40) Yield (c=1, EtOH) ⁱH NMR (ĎMSČ 3 H) CH₃ Val), 0 86 (3H, d, J = 6 8 Hz, CH₃ Val), 1 32 (3H, s, CH₃ A1b), 1 34 (3I δ 0.83 (3H, d, J = 6 8 Hz, m, H-β Val), 3.53 , s, CH₃ A₁b), 1 84-1 98 (1H, (3H, s, OCH₃), 3 85 (1H, dd, J~9 Hz, H- α Val), 5 03 (3H, s, CH₂ Z), 7 10 (1H, d, J = 9 3) Hz, NH Val), 7 27-7 38 (5H, m, C₆H₅ Z), 8 21 (1H, s, NH Alb) Anal calc for $C_{18}H_{26}N_2O_5$ %C = 61 70, $%H = 748,$ % $N = 799$ Found % $C = 6123,$ % $H = 756,$ % N

Z-Aib-Pro-OtBu (8):

With BOP Z -Aib-OH (1 896 g, 8 mmol), H-Pro-OtBu (1 505 g, 8 8 mmol), BOP (3 54 g, 8 mmol) and

DIEA (4 6 ml, 27 mmol) in CH₂Cl₂ (8 ml). Column chromatography (hexane: EtOAc, 40 60)
Recrystallized from MeOH/H₂O Yield 2 65g (82%) M.p.= 105°C Rf = 0 3 (hexane EtOAc, 50 50) [α]_D
= -91° (c=1,EtOH) ¹H NMR (D $%N=7.24$.

With BroP: From Z-Aib-OH (1 mmol). Yield: 84%.

With PyBroP; From Z-Aib-OH (10 mmol), PyBroP (12 mmol). Yield 95%

Boc-Pro-Aib-OMe (9):

With PyBOP, Boc-Pro-OH (0.43 g, 2 mmol), HCl H-Aib-OMe (0.337 g, 2.2 mmol), PyBOP (1.04 g, 2 mmol) and DIEA (1 04 ml, 6 mmol) in CH_2Cl_2 (2 ml). Column chromatography (hexane:EtOAc; 1.1). Yield 88%. M p = 78°C (lt. 18 m p = 80°C); Rf = 0.4 (hexane EtOAc, 50 50) [α]_D= -56° (c=1,EtOH) Anal calc for $C_{15}H_{26}N_2O_5$ %C = 57 31, %H = 8 34, %N = 8 91 Found: %C = 57 14; %H = 8 53, %N $= 8.89.$

With PyBroP, From Boc-Pro-OH (2 mmol) Yield 83% Z-Aib-Ile-OMe (10):

With PyBOP: Z-Aib-OH (0.474 g, 2 mmol), HCl H-Ile-OMe (0 399 g, 2 2 mmol), PyBOP (1 04 g, 2 mmol) and DIEA (1 04 ml, 6 mmol) in CH₂Cl₂ (2 ml) Column chromatography (hexane EtOAc, 50 50). Yield, 80 W M p = 48-50°C. Rf = 0.4 (hexane.EtOAc, 1.1) $[\alpha]_D = +2^\circ$ (c=1,EtOH) ¹H NMR (CDCl₃) δ 0.85
(3H, d, J = 6.8 Hz, CH₃ Ile), 0.87 (3H, t, J = 7.3 Hz), 1.04-1.16 (1H, m, H-y Ile), 1.31-1.43 (1H, m, H-y Ile), 1 50 (3H, s, CH₃ A₁b), 1 53 (3H, s, CH₃ A₁b), 1 81-1 92 (1H, m, H- β Ile), 3 68 (3H, s, OCH₃), 4 53 (1H, dd, $J = 4$ 4 Hz, $J = 8$ 3 Hz, $H - \alpha$ Ile), 5 06, 5 08 (2H, AB, $J = 12$ 7 Hz, CH₂ Z), 5 31 (1H, broad s, NH A1b), 6 81 (1H, broad, NH IIe), 7 25-7 48 (SH, m, C₆H₅ Z) Anal calc for C₁₉H₂₈O₅N₂ %C = 62 64, %H = 7 69, %N = 7 69 Found: %C = 62.91, %H = 7 63, %N

Boc-Ile-Aib-OMe (11)

With PyBOP, Boc-Ile-OH (0 462 g, 2 mmol), HCl H-Aib-OMe (0 338 g, 2 2 mmol), PyBOP (1 04 g, 2 mmol) and DIEA (1 04 ml, 6 mmol) in CH_2Cl_2 (2 ml) Column chromatography (hexane.EtOAc, 1 1) Yield 83% m p = 168°C. $[\alpha]_{p}$ = -28° (c=1, EtOH) Rf = 0.8 (hexane EtOAc, 50 50). ¹H NMR (CDCl₃) δ 0 88 (3H, t, J = 7 3 Hz, CH₃ Ile), 0 90 (3H, d, J = 6 8 Hz, CH₃ Ile), 1 04-1 17 (1H, m, H-γ Ile), 1.42 (9H, s, CH₃ Boc), 1.40-1 50 (1H, H-γ Ile), 1 507 (3H, s, CH₃ Alb), 1 513 (3H, s, CH₃ Alb), 1.77-1 88 (1H, t, Ile), 3.69 (3H, s, OCH₃), 3 85 (1H, ~t, J ~ 7 3 Hz, H- α Ile), 5 00 (1H, broad, NH Ile), 6.49 (1H, s, NH A1b) Anal calc for $C_{16}H_{30}N_2O_5$ %C = 58 18, %H = 9 09, %N = 8 48 Found %C = 58 54, %H = 9 32, %N = 844

 Z -Aib-Aib-OMe (12) :

With PyBOP, Z-Aib-OH (0 237 g, 1 mmol), HCl H-Aib-OMe (0 169 g, 1 1 mmol), PyBOP (0 52 g, 1 mmol) and DIEA (0.52 ml, 3 mmol) in $CH₂Cl₂$ (1 ml) TLC monitoring showed immediate formation of one product (Rf = 0.4, hexane EtOAc, 70.30) which disappeared slowly in favor of 12 Reaction time 16 h Column chromatography (hexane EtOAc, 60 40) Yield 89% M p = 106-108°C (lit 6 m.p = 107-109°C) $Rf = 0.35$ (hexane EtOAc, 60.40)

With PyBroP, From Z-A1b-OH (1 mmol) Reaction time 16 h Yield 77%

With PyBroP/DMAP: Z-Aib-OH (0 4587 g, 1 935 mmol), HCl H-Aib-OMe (0 327 g, 2 13 mmol), PyBroP $(1\ 082\ g, 2\ 32\ mmol = 1\ 2\ eq)$, DIEA (0 8 ml, 2 4 eq), DMAP (0 142 g, 1 161 mmol = 0 6 eq) in CH₂Cl₂ (2 ml) Reaction time 1 h Yield 80%

Boc-Aib-Aib-OMe (13):

With BOP: Boc-Aib-OH (0 203 g, 1 mmol), HCl H-Aib-OMe (0 169 g, 1 1 mmol), BOP (0 442 g, 1 mmol) and DIEA (0.51 ml, 3 mmol) in CH_2Cl_2 (1 ml) Immediate formation of 14 (TLC) which disappeared slowly in favor of 13 Reaction time 16 h Column chromatography (hexane EtOAc, 1 1) Yield 80% M p = 91-92 °C (lit ¹⁸ m p = 86-87 °C) Rf = 0.4 (hexane EtOAc, 1.1)

With PyBOP, From Boc-Aib-OH (2 mmol) Immediate formation of 14 Reaction time 16 h Yield 86% With BroP, From Boc-Aib-OH (1 mmol) Reaction time 24 h Yield 25%

With PyBroP, From Boc-Aib-OH (1 mmol) Reaction time. 24 h Yield 25%

With BroP/DMAP, Boc-Aib-OMe (0 203 g, 1 mmol), HCl H-Aib-OMe (0 169 g, 1 1 mmol), BroP (0 388 g, 1 mmol), DIEA (0.408 ml, 2 4 mmol) and DMAP (0.073 g, 0.6 mmol) in CH₂Cl₂ (1 ml) Reaction time 1

h Column chromatography (hexane.EtOAc, 11) Yield 76%
With PyBroP/DMAP, Boc-Aib-OH (0.322 g, 1.583 mmol), HCl-H-Aib-OMe (0.2673 g, 1.741 mmol),
PyBroP (0.885 g, 1.899 mmol), DIEA (0.65 ml, 3.8 mmol = 2.4 eq.) and DMAP (0.1

Boc-Alb-OBt (14): Boc-Alb-OH (0 203 g, 1 mmol), BOP (0 442 g, 1 mmol) and DIEA (0 34 ml, 2 mmol) in CH₂Cl₂ (1 ml) were stirred at R T TLC (revealed at 254 nm) showed immediate formation of one product. After 30 mm, CH 5%, brine, dried on Na $Cl₂$ was evaporated, the residue was dissolved in EtOAc, washed with KHSO₄ SO_4 , filtered, evaporated. Yield of crude product: 0.24 g (oil). Rf = 0.65 (hexane:EtOAc; 1:1). ¹H NMR (CDCl₃): δ 1.46 (s) and 1 5(s) (9H, Boc); 1 71 (6H, s, CH₃ Aib); 5 35 $(H, broad s, NH Aib);$ 7 3-8.1 (4H, OBt).

Boc-Aib-Aib-OBzl (15):

With BroP/DMAP: Boc-A1b-OH (0 812 g, 4 mmol), Tos A1b-OBzl (1 608 g, 4.4 mmol), BroP (1 552 g, 4 mmol), DIEA (1.63 ml, 9 6 mmol = 2.4 eq) and DMAP (0 293 g, 2 4 mmol = 0 6 eq) in CH₂Cl₂ (4 ml) Column chromatography (hexane EtOAc, 1 1) Yield 65% M $p = 121^{\circ}C$ (lit.¹³: m $p = 121 \cdot 122^{\circ}C$) Rf = 0.6 (hexane.EtOAc, 1:l)

With PyBroP/DMAP: From Boc-Aib-OH (1 mmol) Yield 61% **Z-Aib-Benzacaine (16):**

With BOP, Z-A1b-OH (0.474 g, 2 mmol), Benzocame (0 363 g, 2 2 mmol), BOP (0.884 g, 2 mmol) and DIEA (0.68 ml, 4 mmol) in CH_2Cl_2 (2 ml) TLC monitoring showed immediate formation of one product $(Rf = 0.4$, hexane EtOAc, 70:30) which reacted slowly to give 16 Reacuon ume. 24 h. Work-up as general procedure Yield of crude product: 1.1 g TLC still showed the same product, benzocaine and 16 ¹H NMR (CDCl₃). about 30% of 16; numerous signals at $\delta = 72-8$ ppm revealed the presence of Z-A1b oxybenxotnazolyl ester

With BroP/DMAP, Z-Atb-OH $(3.92 \text{ g}, 16 \text{ f}, \text{mmol})$, Benzocatne $(2.48 \text{ g}, 15 \text{ mmol})$, BroP $(6.40 \text{ g}, 16.5 \text{ m}$ mmol), DIEA (2.8 ml, 16.5 mmol) and DMAP (2 02 g, 16 5 mmol) in CH_2Cl_2 Column chromatography (hexane. EtOAc; 60:40). Yield 66% M p.= 12 (15 ml) Reactton tune' 2 h 6-7°C (ht.^{24} : m p = 131.0-131.2°C). Rf = 0.4 (hexane:EtOAc; 60:40).

With PyBroP/DMAP, From Z-Aib-OH (1 mmol). Reaction time-2h Yield. 65%

Z-L-Leu-Aib-Pro-OtBu (eq.1):

 $Tfa.H-Aib-OAllvl(I7):$

a) *Boc-Aib-O-Allyl* : Boc-Aib-OH (6 7 g, 33 mmmol) were dissolved in DMF (15 ml); allyl bromide (2 86 ml, 33 mmol) and Cesium carbonate (5 376 g, 16.5 mmol) were successtvely added and the reaction mixture stirred for 2 h at room temperature CsBr was filtered and the solvent removed *in vacua. The* crude product was dissolved in CH_2Cl_2 , washed with water, dried on Na₂ MeOH/H₂O Yield 79% M.p = 50-51°C Rf = 07 (hexane EtOA $SO₄$, evaporated and recrystallized from c, 70⁻30) ^IH NMR (DMSO (6H, s, CH₃ A1b), 1 35 (9H, s, CH₃ Boc), 4 51 (2H, d, J = 5 4 Hz, O-C<u>H₂</u>-CH=CH₂), 5 18 (1H, d, J = 10.8 Hz, O-CH₂-CH=C<u>H</u>₂), 5 30 (1H, dd, J = 17.1 Hz, J = 1 9 Hz, O-CH₂-CH=C<u>H₂), 5.87 (1H, m, O-CH₂-</u> $CH=CH_2$, 7 22 (1H, broad s, NH Atb) Anal calc for $C_{12}H_{21}NO_4$ % $\bar{C} = 5924$; %H = 8.70, %N \bar{C} Found: $\%C = 5922$; $\%H = 8.71$; $\%N = 5.83$

b) Tfa-Atb-O-Allyl Boc-Aib-OAllyl (6 g, 24 69 mmol) was treated with trifluoroacetic acid (12 35 ml) for 35 minutes Excess of TFA was removed in vacuo and the product crystallized from ether/hexane (6H, s, CH_3 Aib), 4.71 (2H, dt, J = 5 9 Hz, J = Yield 88% M p = 65-68°C. ¹H NMR (DMSO d₆) δ 1 48 (5.36 (lH, dt. J = 19 0 1 4 Hz, O-CH₂-CH=CH₂ 5.28 (1H, dt, J = 11 0 Hz, J = 1.2 Hz, O-CH₂-CH=C<u>H</u> 5 94 (1H, m, O-CH₂-C<u>H</u>=CH₂), 8 51 (3H, NH ⁺ A_{tb}), Anal. calc for $C_9H_{14}NO_4F_3$ %C = 42.02; %H = 5.45; %N = 5.45. Found: %C = 42.00, %H = 5.50; %N = 5.23 1) Epimerization study with PyBOP

Z-L-Leu-Aib-O-Allyl (18): 17 (3.115 g, 12.1 mmol), Z-L-Leu-OH (3.57 g, 13.4 mmol), PyBOP (6.41 g, 12.3 mmol), DIEA (6.2 ml, 36.3 mmol) in CH₂Cl₂ (12 ml) Column chromatography (hexane EtOAc; 70:30). Yield: 87%. Rf = 0.3 (hexane EtOAc; 70.30) M p. = 51-52 °C. [α]_D = -25⁸ ¹H NMR (DMSO d₆): δ 0.85 (3H, d, J = 6.8 Hz, CH₃ Leu), 0.87 (3H, d, J = 6.8 Hz, CH₃ Leu), 1 34 (3H, s, CH₃ Ab), 137 (3H, s, CH₃ Ab), 132-1.38 (2H, m, H-β Leu), 1.62 (1H, m, H-γ Leu), 4.06 (1H, ~q, J -
8.8 Hz, H-α Leu), 4.46 (2H, d, J = 4 9 Hz, O-CH₂-CH=CH₂), 5 02 (2H, s, CH₂ Z), 5.15 (1H, σ q, J -
0-CH₂-CH 64 60; %H = 7 74, %N = 7.17, Found %C = 64 69, %H = 7.72, %N = 7 13.

Z-L-Leu-Aib-OH (19): 18 (4.1 g, 10 5 mmol) and morphohn (2 75 ml, 31.5 mmoi) m THF (200 ml) were stured for 5 minutes in the presence of tetrakis-(mphenylphosphine)palladium(0) (1 21 g, 1 05 mmol) according to 33 THF was evaporated, the yellow product dissolved in EtOAc, washed 3 x with 5% KHSO₄, extracted with 5% NaHCO₃, the aqueous solution was acidified with 6N HCl and extracted with CH₂Cl₂ The product was crystallized from Et_2O/h exane Yield 63% ⁱH NMR (DMSC M p =113-115°C $[\alpha]_{D}$ = -25° (c=1, EtOH) (3H, s, $CH₃$ Aib), $\bar{1}$) δ 0.85 (3H, d, J = 6 4 Hz, CH₃ Leu), 0 87 (3H, d, J = 6.3 Hz, CH₃ Leu), 1 32 34 (3H, s, CH₃ Aib), 1 37-1 46 (2H, m, H-β Leu), 1 54-1.71 (1H, m, H-γ Leu), 4 05 (1H, ~q, J ~ 8 8 Hz, H- α Leu), 5 02 (2H, s, CH₂ Z), 7 26 (1H, d, J = 8 3 Hz, NH Leu), 7.28-7 40 (5H, m, $=7$ C_6H_5 Z), 7 97 (1H, s, NH Aib), 12.1 (1H, broad s, COOH). Anal calc for $C_{18}H_{26}N_2O_5$ %C = 61 70, %H = 7 98 Found %C = 61 39; %H = 7 26, %N = 8.08.

Z-L-Leu-Aib-Pro-OtBu (20): 19 (0 175 g, 0 5 mmol), HCl H-Pro-OtBu (0 114 g, 0 55 mmol), PyBOP (0.26 g, 0 5 mmol) and DIEÁ (0.264 ml, 1.5 mmol) in CH₂Cl₂ (0 5 ml) Reaction time: 2 h. Yield 96% The crude product contained no D-Leu diastereoisomer 24 (Yield and epimerization were studied by HPLC see below)

2) Epimerization study with BroP:

Z-L-Leu-Aib-OAllyl (18): 17 (5.02 g, 20 mmol), Z-L-Leu-OH (5 83 g, 22 mmol), BroP (7 76 g, 20 mmol) and DIEA (10.2 ml, 60 mmol) in CH₂Cl₂ (20 ml) Column chromatography (hexane EtOAc, 70:30) Yield 74% M p = 49-52 °C. $[\alpha]_{D}$ = -25° (c=1, EtOH)

Z-L-Leu-Aib-OH (19): 18 (3.0 g, 7 7 mmol), morpholm (2 01 ml, 23 1 mmol) in THF (150 ml) and tetrakis-(triphenylphosphine)palladium(0) (092 g, 08 mmol) according to ³³ Yield 74%. M p = 115°C $[\alpha]_{D} = -25^{\circ}$ (c=1, EtOH)

Z-L-Leu-Aib-Pro-OtBu (20): From 19 (0.5 mmol) with PyBOP Reaction time 1 h Yield 81 % % diastereoisomer 24: 1 5 % (Yield and epimerization were studied by HPLC see below)

Z-L-Leu-Aib-Pro-OtBu (eq.2):

HCl.H-Aib-Pro-OtBu (21): 8 (25 g, 64 mmol) from PyBroP, Pd/C (05 g) and concentrated HCl (0.534 ml) in MeOH (60 ml) were stirred under H₂ for 1 h 45 After filtration, MeOH was evaporated and the product crystallized from MeOH/ether. Yield. 87%. M p = 160°C (decomposition) $[\alpha]_{D}$ = -77° (c=1, EtOH) ¹H NMR (DMSO d₆). δ 1 38 (9H, s, CH₃ tBu), 1 56 (3H, s, CH₃ Aib), 1 59 (3H, s, CH₃ Aib), 1 72-1 82 (1H, m, Pro), 1.84-1 98 (2H, m, Pro), 2 07-2 17 (1H, m, Pro), 3 65 (2H, m, H- δ Pro), 4 25 (1H, m, H-α Pro), 8 41 (3H, s, NH₃⁺ Atb) Anal calc for C₁₃H₂₄N₂O₃Cl² %C = 53.51, %H = 8 29, %N = 9 60,
%Cl = 12 15 Found. %C = 53.71, %H = 8 59, %N = 9 56, %Cl = 11 87

1) Epimerization study with PyBOP:

Z-L-Leu-Aib-Pro-OtBu (20): 21 (0 585 g, 2 mmol), Z-L-Leu-OH (0 584 g, 2 2 mmol), PyBOP (1 144 g, 2 mmol) and DIEA (1 02 ml, 6 mmol) in CH₂Cl₂ (2 ml) The crude product contained no Z-D-Leu-Aub-Pro-OtBu 24 (HPLC: see below) Column chromatography (hexane EtOAc, 1:1) Yield: 81%
M p = 150-152°C [α]_D = -89° (c=0 7, EtOH) Rf =0 40 (hexane EtOAc, 50.50) (lit ¹¹: M p = 144-145°C,
[α]_D = -82 2 (c=0.7 3 49 (2H, m, H-δ Pro), 4.07 (2H, m, H-α Pro, H-α Leu), 4 97, 5 06 (2H, AB, J = 12.6 Hz, CH₂Z), 7 26-7 46 (6H, m, C₆H₅Z, NH Leu), 8 24 (1H, s, NH Alb)

2) Epimerization study with PyBroP:
Z-L-Leu-Aib-Pro-OtBu (20): 21 (0 585 g, 2 mmol), Z-Leu-OH (0 584 g, 2 2 mmol), PyBroP (1 03 g, 2.2 mmol) and DIEA (1.02 ml, 6 mmol) in CH₂Cl₂ (2 ml) The crude product contained 1% Z-D-Leu-Aib-Pro-OtBu 24 (for HPLC assay see below) Column chromatography (hexane EtOAc, 1 1) Yield 77% M p = 149-151°C [α]_D = -89° (c=0.7, EtOH)

Z-D-Leu-Aib-Pro-OtBu (24) (eq.3):

Z-D-Leu-Aib-OMe (22): Z-D-Leu-OH (4 1 g, 15 47 mmol), HCl H-Aib-OMe (3 4 g, 22 mmol), BroP (7.76 g, 20 mmol) and DIEA (10.2 ml, 60 mmol) in CH₂Cl₂ (20 ml) Column chromatography
(hexane.EtOAc, 70 30) Yield 75% M p = 86-87°C [α]_D = +25° (c=1, EtOH) Anal calc for
C₁₉H₂₈N₂O_S %C = 62 64, %H = 7

ml) for 1h The reaction mixture was then solved in EtOAc (100 ml), the carboxylate extracted with H₂O, acidified with 1N HCl and finally extracted with CH₂Cl₂ Yield 72% M p = 110-111°C [α]_D = +25° $(c=1, EtOH)$

Z-D-Leu-Aib-Pro-OtBu (24): 23 (0 35 g, 1 mmol), H-Pro-OtBu (0 197 g, 1 1 mmol), BroP (0 388 g, 1 mmol) and DIEA (0.34 ml, 2 mmol) in CH_2Cl_2 (1 ml) Reaction time 1 h. The crude product contained 1 3% Z-L-Leu-Aib-Pro-OtBu 20 (HPLC see below) Column chromatography (hexane EtOAc, 60 40) Yield 70% The product still contained 0.3% D-isomer after purification $\mathbf{M} \mathbf{p} = 150 \cdot 151^{\circ} \text{C}$ [α]_D= -36° (c=1, EtOH) Rf = 0 36 (hexane EtOAc, 50 50) ¹H NMR (DMSO d₆) δ 0 85 (3H, d, J = 7 4 Hz, CH₃
Leu), 0 87 (3H, d, J = 6 6 Hz, CH₃ Leu), 1 32 (3H, s, CH₃ Aib), 1 34 (3H, s, CH₃ Aib), 1 36 (9H, s, tBu),
1 30-Leu), 7 95 (1H, s, NH Aib) Anal calc for $C_{27}H_{42}N_3O_6$ %C = 64 26, %H = 8 39, %N= 8 33 Found %C = 64 58, %H = 8 58, %N = 8 34.

Z-Ile-Aib-Benzocaine (eq.4):

HCl.H-Aib-Benzocaine (25): 16 (1921 g, 5 mmol), concentrated HCl (0 625 ml, 7 5 mmol) and Pd-C (150 mg) in MeOH (40 ml) were allowed to react with $H₂$ The product was recrystallized from

MeOH/Et₂O Yield. 90% M p. = 146-150°C (decomposition) ¹H NMR (CDCl₃): δ 1.31 (3H, t, J = 69 Hz, **COLLET THE COLLET THE COLLET THE COLLET COLLET THE COLLET TO ATT AID, 1.66 (6H, s, CH₃ Aib), 4 29 (2H, q, J = 6 9 Hz, OCH₂CH₃), 7 88, 7 95 (4H arom, AA'BB', J = 8 1 Hz), 8 49 (3H, broad s, NH₃⁺ Aib), 10 56 (1H,**

With BroP, Z-Be-OH (0 265 g, 1 mmol), 25 (0 321 g, 1 1 mmol), BroP (0 388 g, 1 mmol) and DlEA (0 51 ml, 3 mmol) in CH₂Cl₂ (1 ml) Reaction time: 1 h The crude product contained \sim 0 5% Z-D-allo-Ile-Ail Benzocame (NMR) Column chromatography (hexane EtOAc,1 1) Yield 60% M p = 79-80 °C $[\alpha]_{D}$ = +46° (c=1 1,EtOH) (lit.¹¹ m.p.= 78 8-79 5⁸C. [α]_D = 46 4° (c=1 1, EtOH)) Rf = 0 5 (hexane.EtOAc, 1 1) ¹H NMR (CDCl₃) showed⁻ H- α Be (δ 3 81, t, J = 6 5 Hz) and 0 5% H- α D-allo-Ile (δ 3 95) (see ²⁸)

With PyBroP, Z-Ile-OH (0.3076 g, 1.159 mmol), 25 (0 3008 g, 1 049 mmol), PyBroP (0 7458 g, 1 6 mmol) and DIEA (0.54 ml, 3.15 mmol) in CH_2Cl_2 (1.05 ml) Reaction time 1 h The crude product contained no D-diastereoisomer (NMR). Column chromatography (hexane EtOAc, 60 40) Yield 60% M p = 77-79°C $[\alpha]_{\text{D}} = +47^{\circ}$ (c=1 1, EtOH) Rf = 0.4 (hexane : EtOAc , 60 40) ¹H NMR (CDCl₃) showed H- α Ile (δ 3 81, t, $J = 65$ Hz) and no signal corresponding to D-allo-Ile isomer (δ 3.95 ppm) (see ²⁸) The recording conditions made it possible to confirm that D-isomer was $< 0.2\%$

Z-L-Leu-Aib-Pro-OtBu (20): Fragment coupling (table 3):

Z-L-Leu-Aib-OH 19 used for this study was synthesized from PyBOP (see above) Yields and $%$ epimer were measured on the reaction mixture without further work-up

1) Epimerization: The peaks corresponding to the diastereoisomers 20 and 24 were well separated (R $T =$ 5 7 and 6 5 min respectively) using the following conditions

- normal phase column Ultrasphere S₁ 5 μ 250 x 4 6 mm (Beckman)

 -2 ml/mm flow of hexane EtOAc, 1 1 ($isocrate$)

- vlsuahsatlon at 254 nm Integration using Beckman system Gold program

2) Yield: The yields of 20 were determined using the commercially avalable Z-Pro-Leu-OEt as internal reference under the following conditions:

- Ultrabase $C85\mu 150 \times 46$ mm column (SFCC)

- vlsuahzahon at 214 nm

 $-$ 2 ml/mm flow of acetonitrile water TFA, 50 50 $1^{\circ}/\circ$

Z-Pro-Leu-OEt $R T = 39$ min, Z-Leu-Aib-Pro-OtBu 20 (and 24) $R T = 48$ min

Coupling with PyBOP, this reaction is described above (eq 1) Reaction time 2 h Yield 96% No epimenzanon

With PvBroP; **19 (0 175 g, 0 5 mmol),** HCl H-Pro-OtBu (0 114 g, 0 55 mmol), PyBrOP (0 233 g, 0 5 mmol) and DIEA (0 26 ml, 1 5 mmol) in CH₂Cl₂ (0 5 ml) Reaction time 2 h Yield 74% No epimerization

<u>With PyBroP/DMAP,</u> 19 (0 175 g, 0 5 mmol), HCl H-Pro-OtBu (0 114 g, 0 55 mmol), PyBroP (0 233 g, 0 5 mmol), DIEA (0 237 ml, 1 4 mmol) and DMAP (0 189 g, 0 14 mmol) in CH₂Cl₂ (0 5 ml) Reaction time 1 h Yield 79% No epimerization

With **DCC, 19 (0 175 g, 0 5 mmol),** HCl H-Pro-OtBu (0 114 g, 0 55 mmol), DCC (0 104 g, 0 5 mmol), DIEA (0 264 ml, 1 5 mmol) in CH_2Cl_2 (0 5 ml) Reaction time 5 min at 0 °C, 2 h at RT Yield 86% No epimerization

With DCC/HOBt, 19 (0 175 g, 0 5 mmol), HCl H-Pro-OtBu (0 114 g, 0 55 mmol), DCC (0 104 g, 0 5 mmol), HOBt $(20\% H_2O)$ (0 0822 g, 0 5 mmol), DIEA (0 264 ml, 1 5 mmol) in CH₂Cl₂ (0 5 ml) Reaction time. 5 min at 0° C, 2 h at RT Yield 85% No epimenzation

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References and notes

1-Abbrevlatlons and symbols follow the recommendations of the IUPAC-IUB Joint Commlsslon on Biochermical Nomenclature (Eur J Biochem 1984, 138, 9) In addition the following abbreviations are used **BQP:**
PvROP®: *(* **PyBOP** (1H-1, 2, 3-benzotriazol-1-yloxy)-tris(dimethylamino)-phosphonium hexafluorophosphate : (lH- 1, 2, 3-benzomazol-1-yloxy)-ms(pyrrohdmo)-phosphomum hexafluorophosphate, **BroP:** bromo-tris(dimethylamino)-phosphonium hexafluorophosphate. PyBroP: bromo-tris(pyrrolidino)phosphonlum hexafluorophosphate, DCC: dlcyclohexyl-carbodnmlde, **BOP-Cl: bls(2-oxo-3-** oxazolidinyl)phosphinic chloride, HOBt: 1, 2, 3-benzotriazole-1-hydroxyde, DIEA: diisopropylethylamine; DMAP: 4-dimethylaminopyridine, CDI: carbonyl diimidazole, MA: mixed anhydride, AE active ester; Aib: α -aminoisobutyric acid

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